

Le stockage électrochimique : Les batteries à ions multivalents

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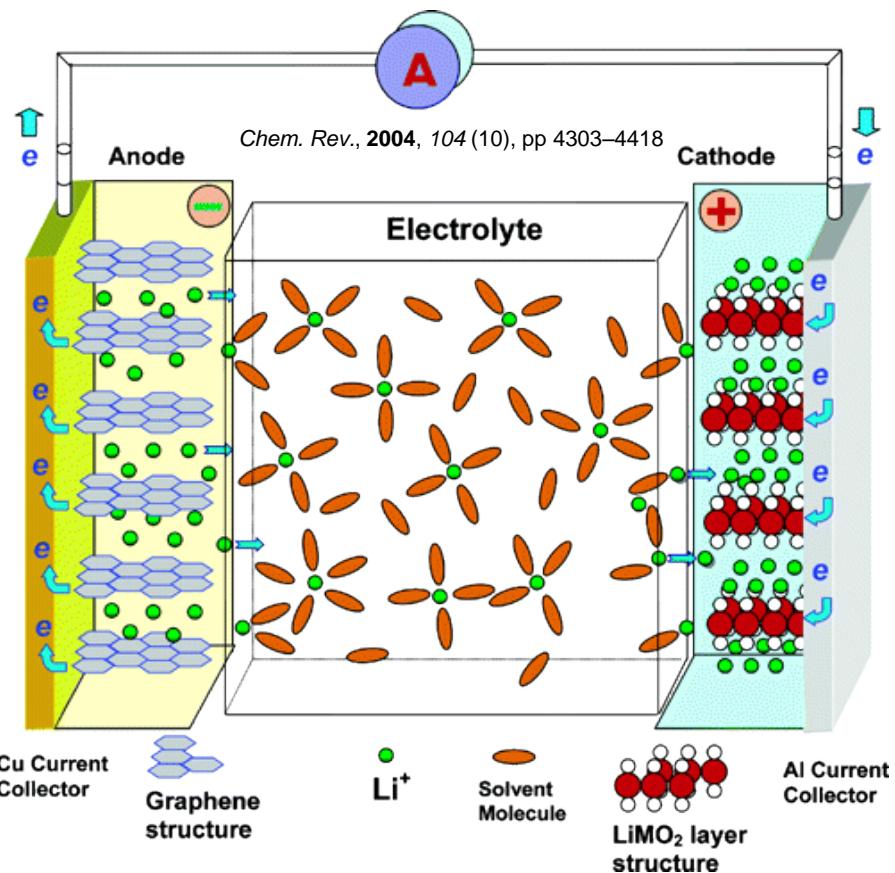
^e **Univ Orleans, CNRS UPR CEMHTI 3079, Orleans, France**

^f **X-ray Science Division, Advanced Photon Source, Argonne National Laboratory, 9700 South Cass Avenue, Argonne, Illinois 60439, United States**

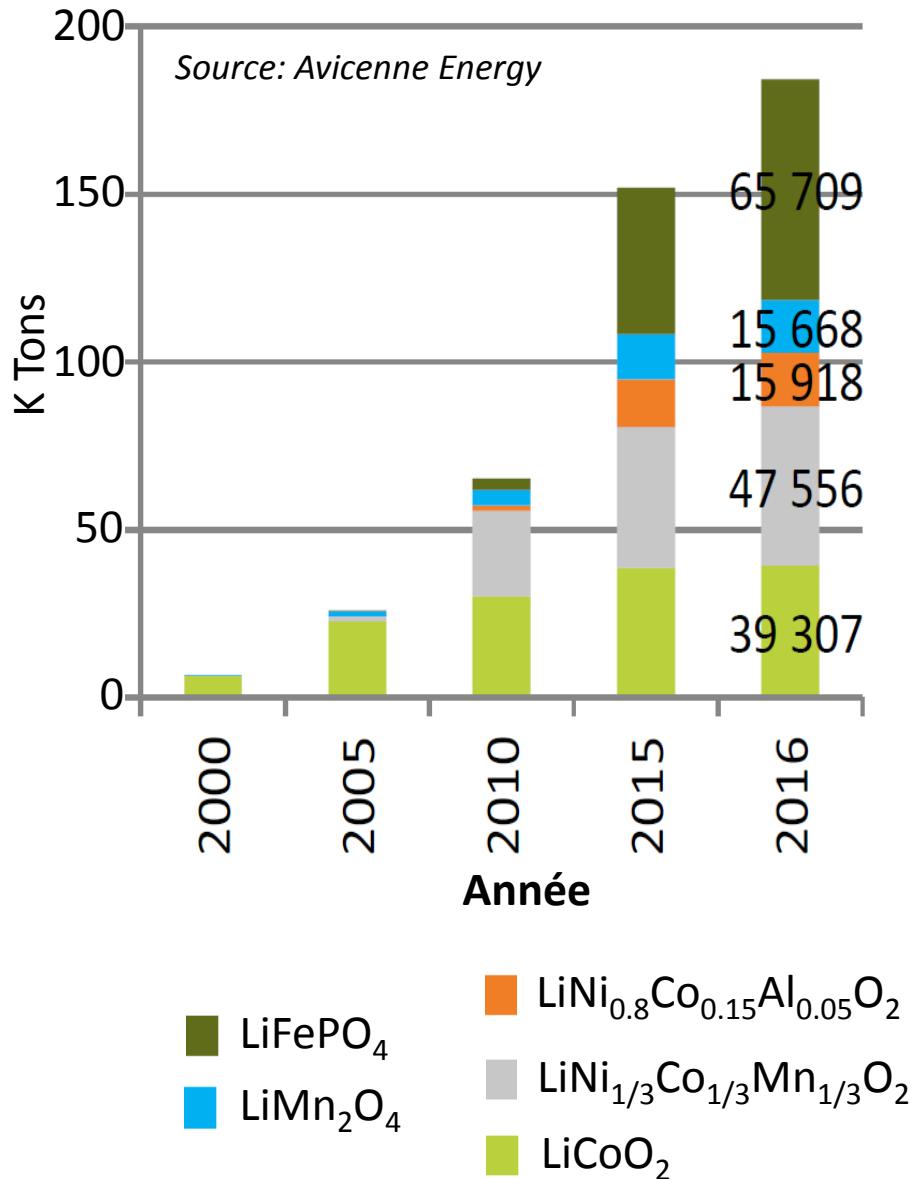
^g **Réseau sur le Stockage Electrochimique de l'Energie (RS2E), FR CNRS 3459, 80039 Amiens, France.**

Les batteries lithium-ion

Le principe



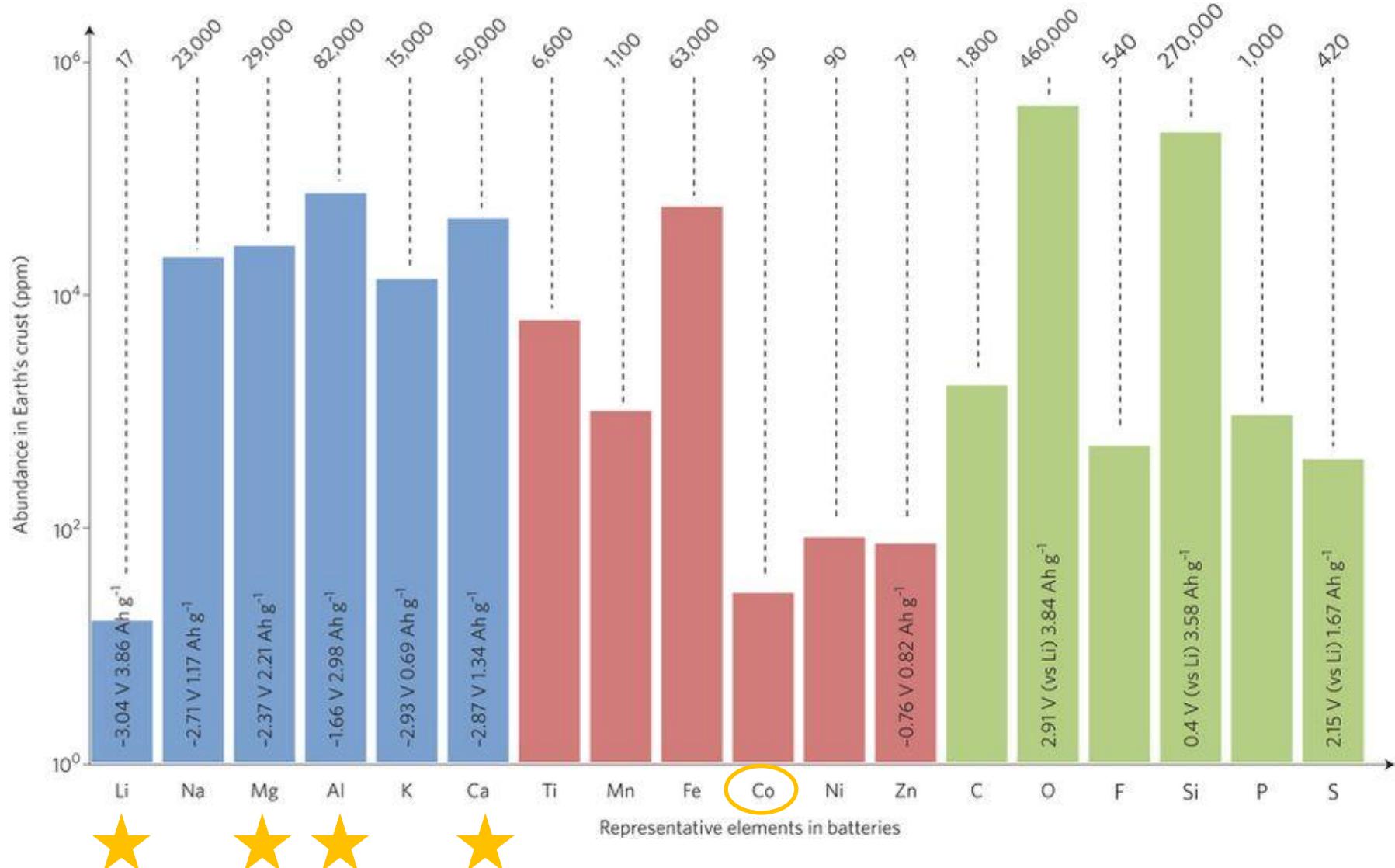
La production des matériaux cathodes



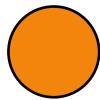
Les applications :

- Dispositifs électroniques (tablettes, portables, objets connectés), etc.
- Transports (vélos, véhicules électriques)
- Le stockage stationnaire

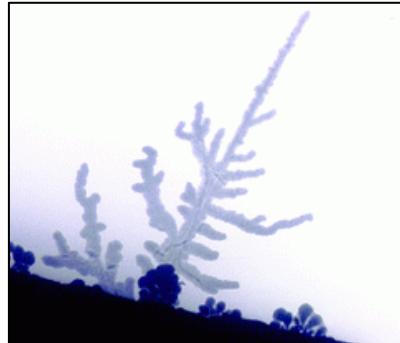
Au-delà du lithium-ion



Les batteries à ions multivalents

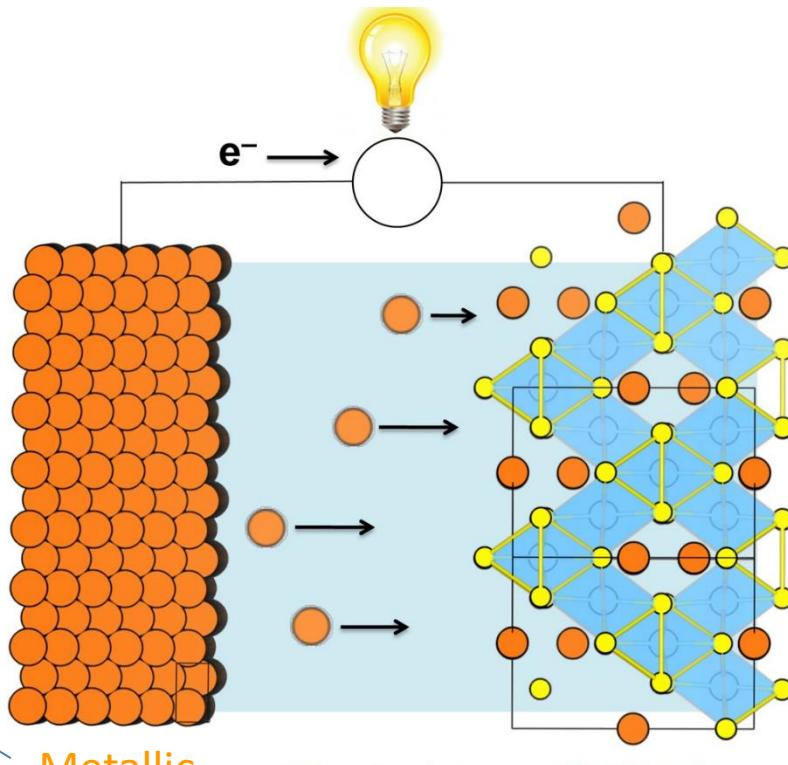


Mg, Al, Ca, Zn
✓ Abondants



✓ Pas de formation de dendrites

✓ Densité d'énergie par Volume

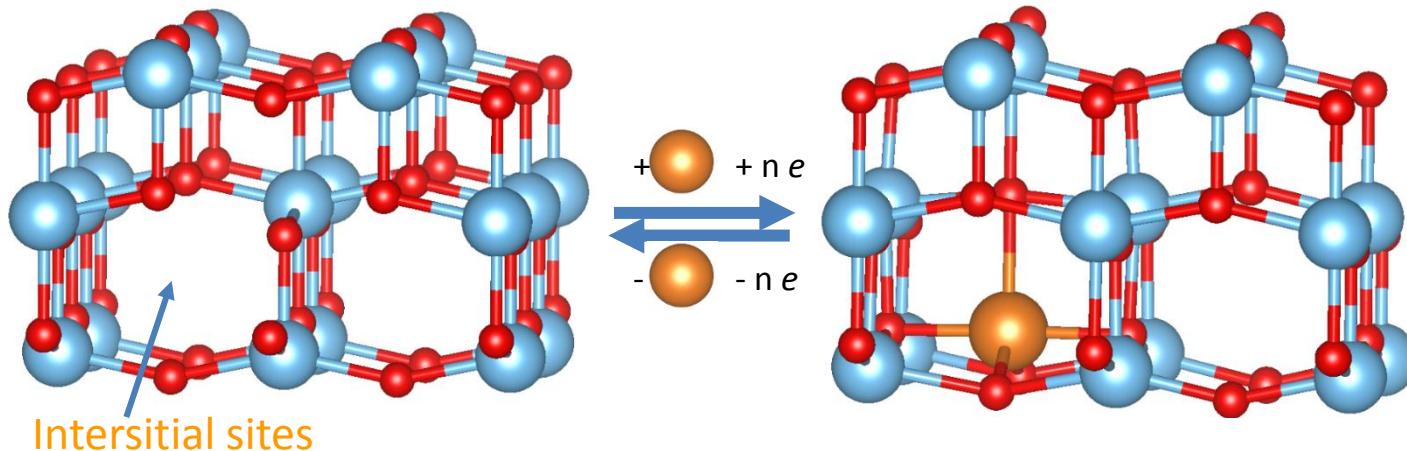


✗ Conductivité ionique
Dépôt/dissolution
Stabilité électrochimique

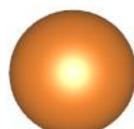
✗ Peu de matériaux électrochimiquement actifs vis-à-vis des ions MVs

La chimie d'intercalation des ions multivalents

TiO₂



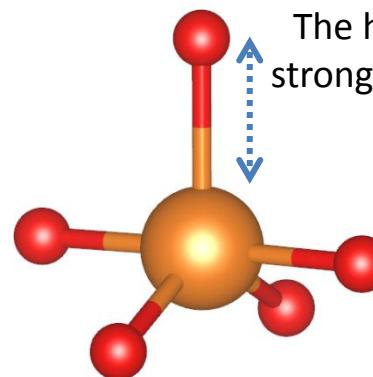
	Li ⁺	Na ⁺	Mg ²⁺	Al ³⁺
Shannon's ionic radii (Å) (CN=VI)	0.76	1.02	0.72	0.53
Polarizing power (Z/r ²)	1.73	0.96	3.85	10.68



= Li⁺ Mg²⁺ Al³⁺

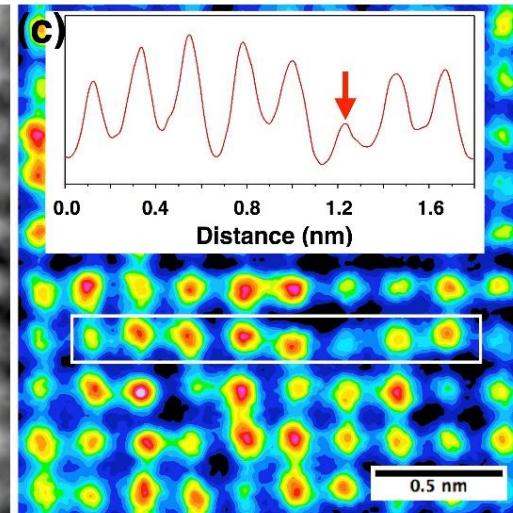
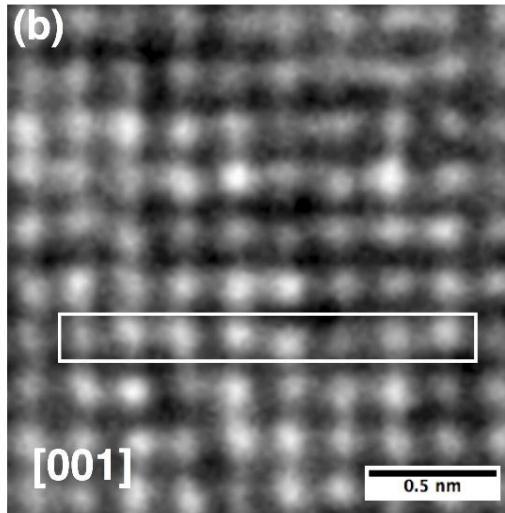
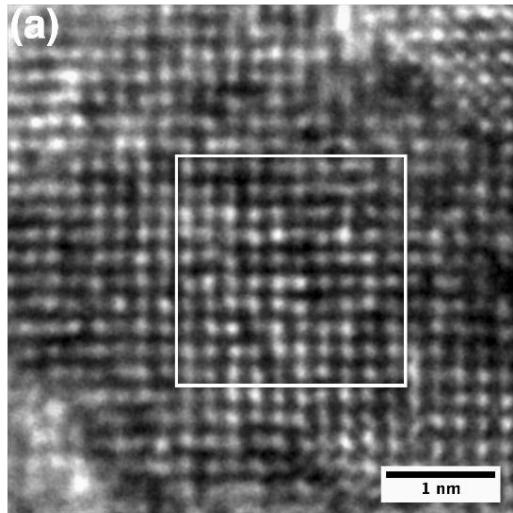
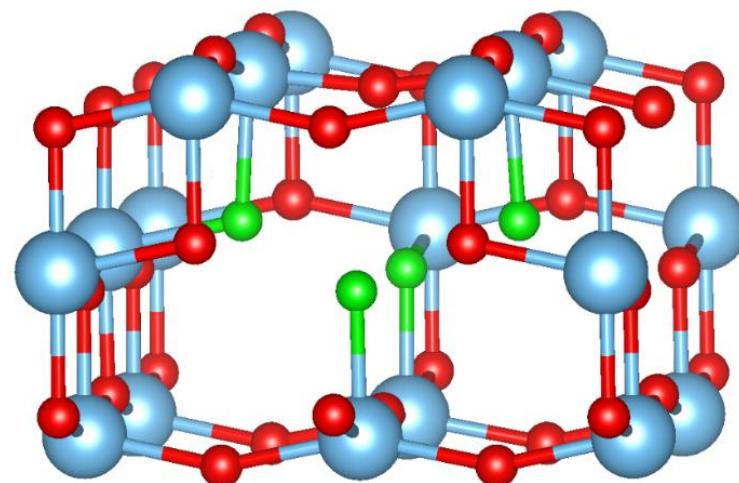
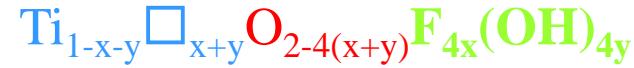
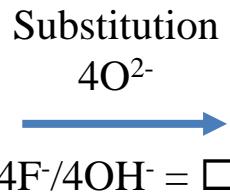
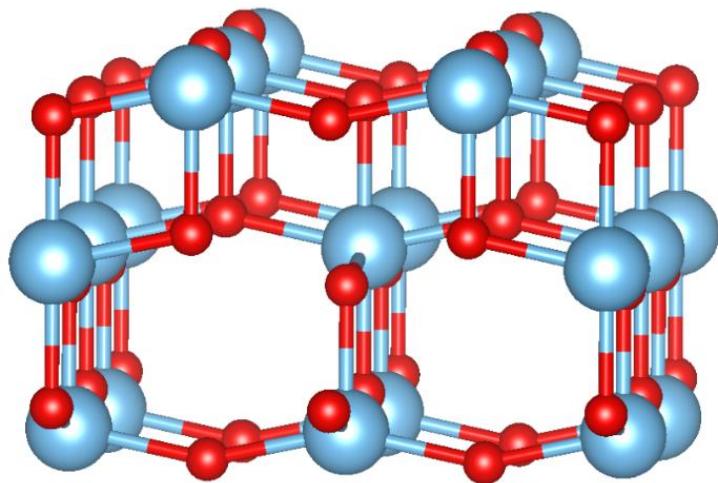


???



The higher the cation's charge, the
stronger the electrostatic interaction,
AND
the weaker the mobility

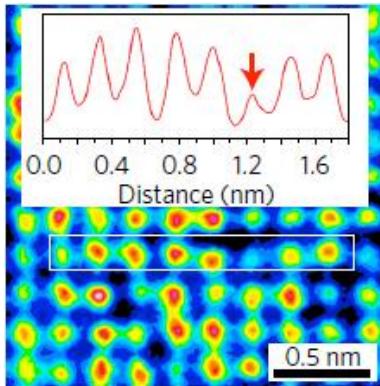
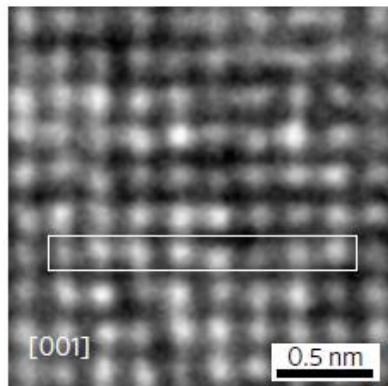
La chimie des défauts



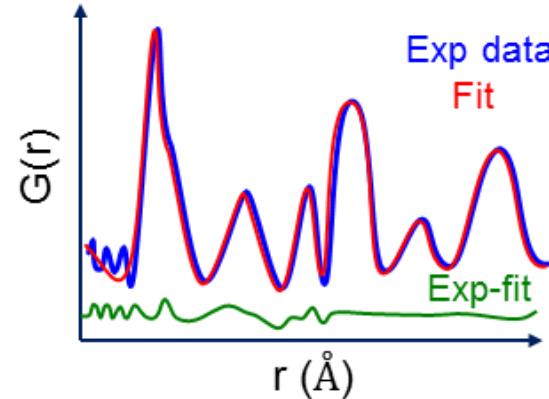
Des matériaux complexes



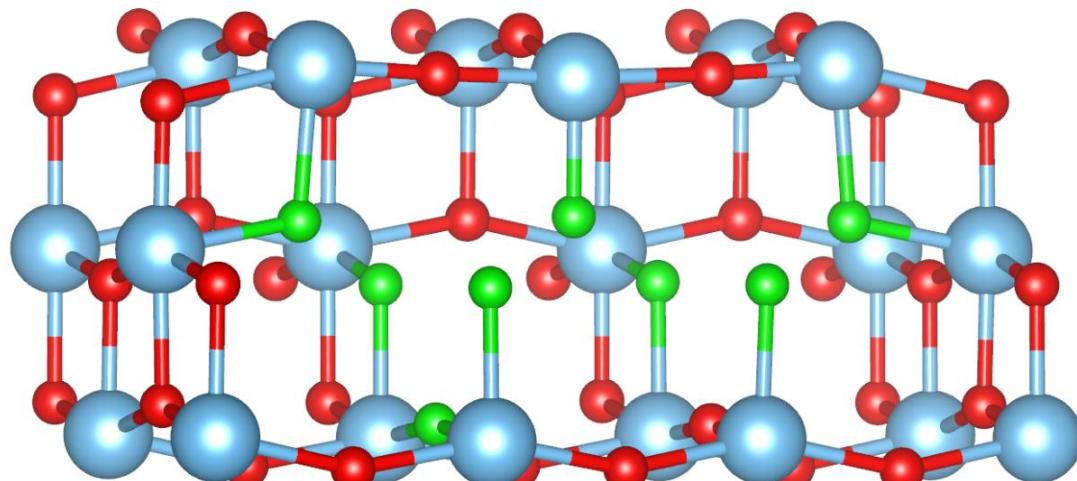
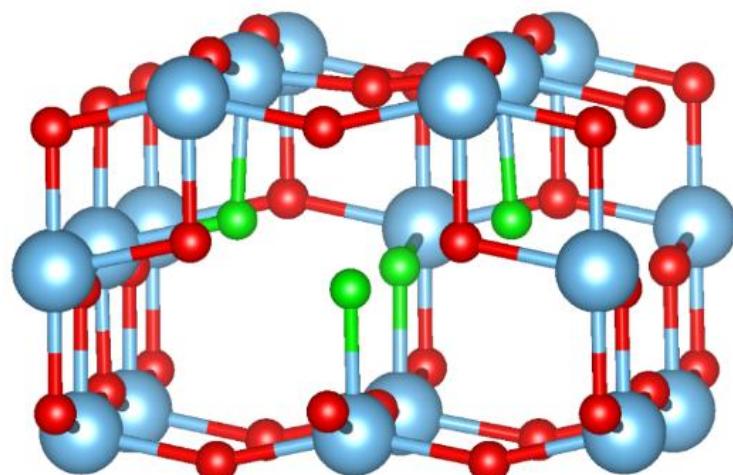
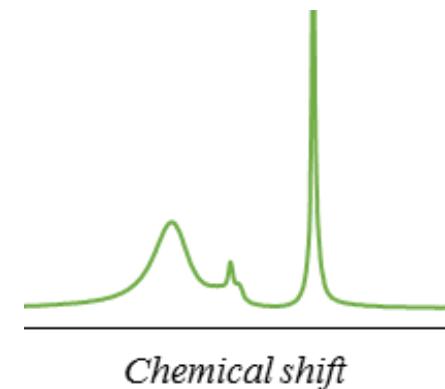
High-resolution TEM



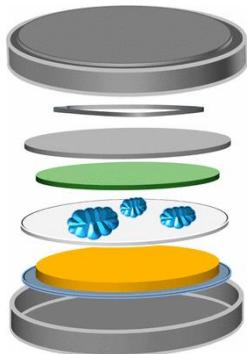
X-ray Pair Distribution Fonction



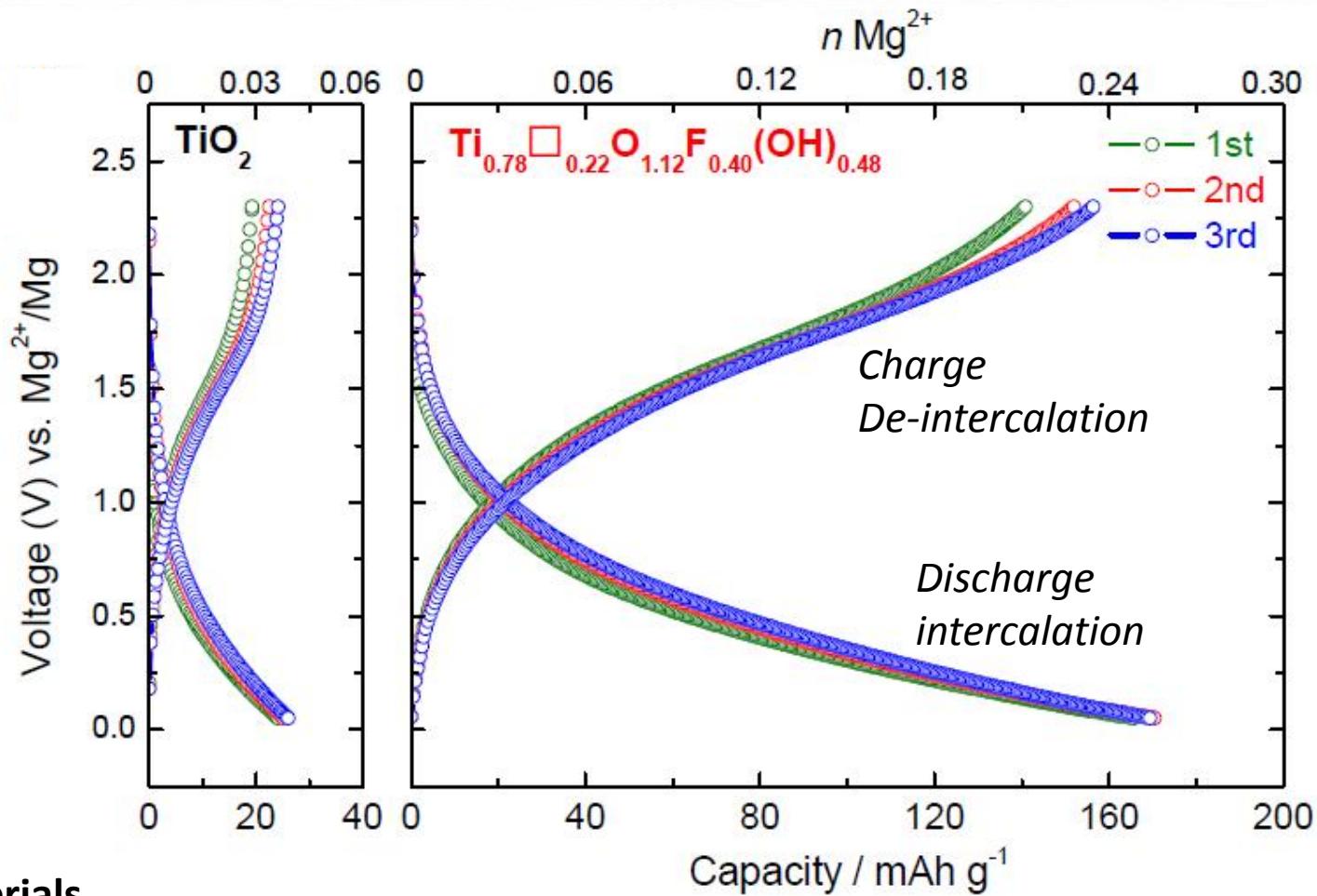
$^{19}\text{F}, ^1\text{H}$ Solid-state NMR



Activité électrochimique vs. Mg²⁺/Mg



(-) Mg || 0.2 mol L⁻¹ MgPhCl₂/AlCl₃ in THF electrolyte || F-TiO₂ (+)



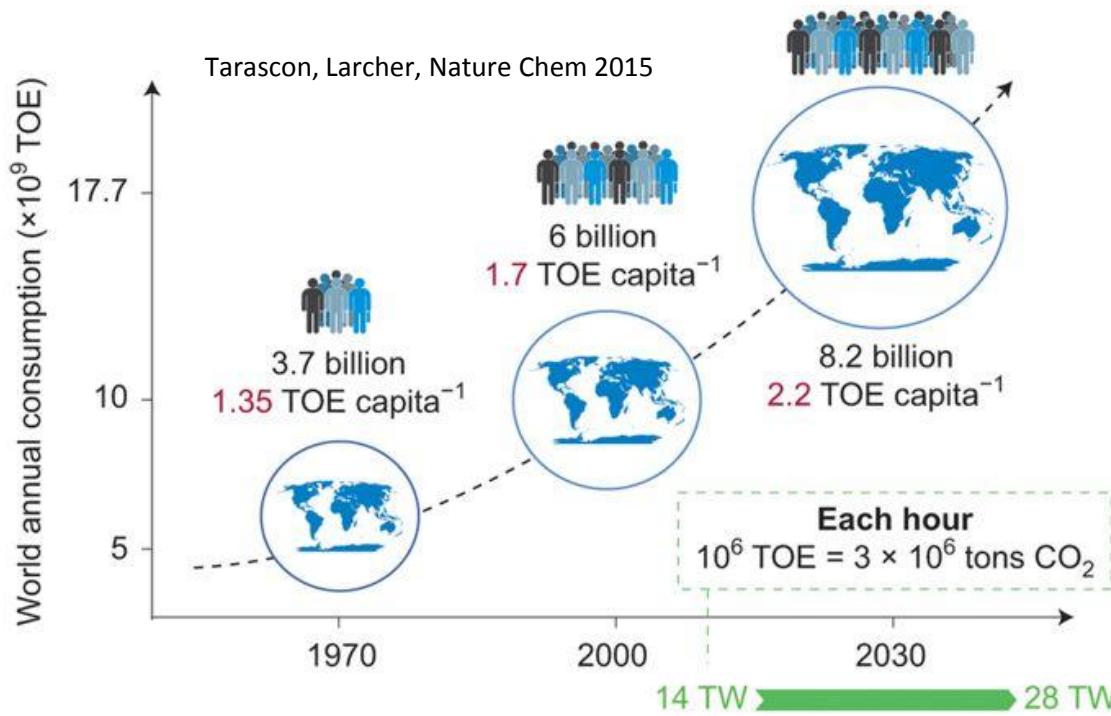
Nature Materials
16 (11), 1142, 2017

Effet des lacunes sur l'activité électrochimique

Conclusion

- Les enjeux du stockage électrochimique :

- (i) Développement de systèmes capables la demande croissante,



- (ii) Développement de nouveaux matériaux (à forts taux de recyclabilité)

Remerciements

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Christophe Legein, Franck Fayon, Olaf Borkiewicz



FRENCH FLUORINE NETWORK



French research network on
electrochemical energy storage



Société Chimique de France

References

High Substitution Rate in TiO₂ Anatase Nanoparticles with Cationic Vacancies for Fast Lithium Storage

Wei Li, Dario Corradini, Monique Body, Christophe Legein, Mathieu Salanne, Karena W. Chapman, Peter J. Chupas, Anne-Laure Rollet, Christian Julien, Karim Zhabib, Mathieu Duttine, Alain Demourgues, Henri Groult, and Damien Dambournet

Chemistry of materials, 27 (14), p 5014, 2015.

Reversible Magnesium and Aluminium-ions Insertion in Cation-Deficient Anatase TiO₂

Toshinari Koketsu, Jiwei Ma, Benjamin J. Morgan, Monique Body, Christophe Legein, Walid Dachraoui, Mattia Giannini, Arnaud Demortiere, Mathieu Salanne, François Dardoize, Henri Groult, Olaf J. Borkiewicz, Karena W. Chapman, Peter Strasser, and Damien Dambournet

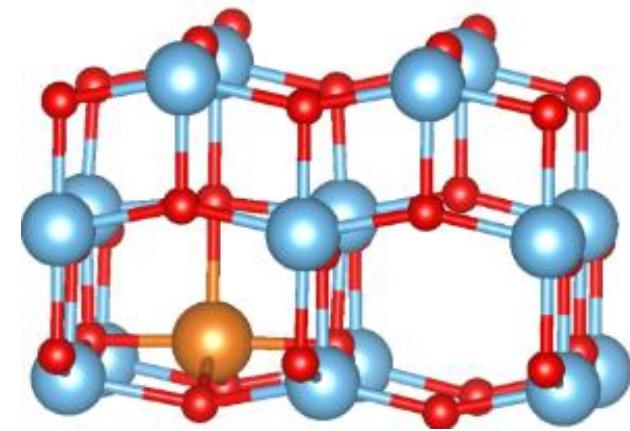
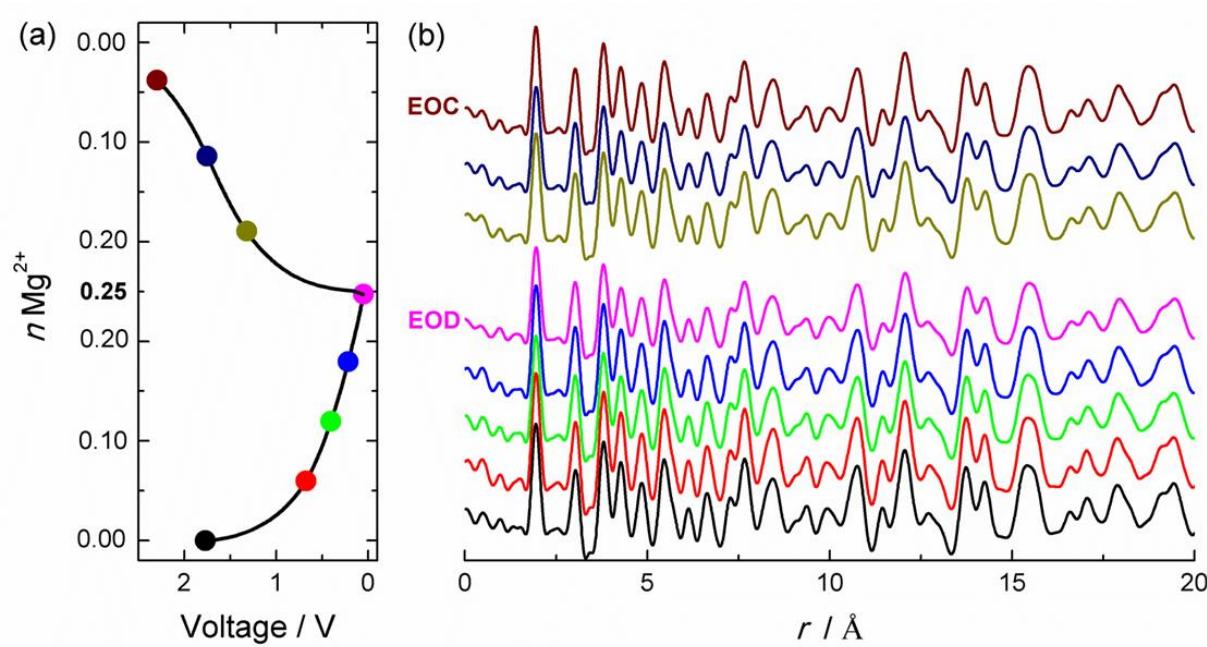
Nature Materials, 16, p 1142, 2017.

La Chimie des défauts, un outil pour le stockage de l'énergie

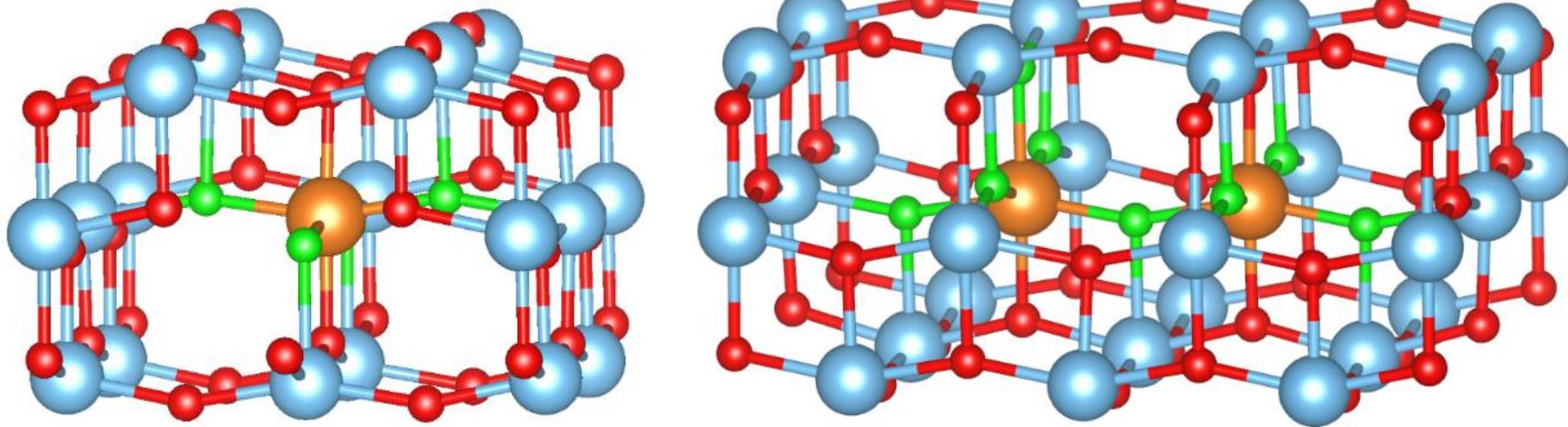
D. Dambournet

L'actualité chimique, 2018.

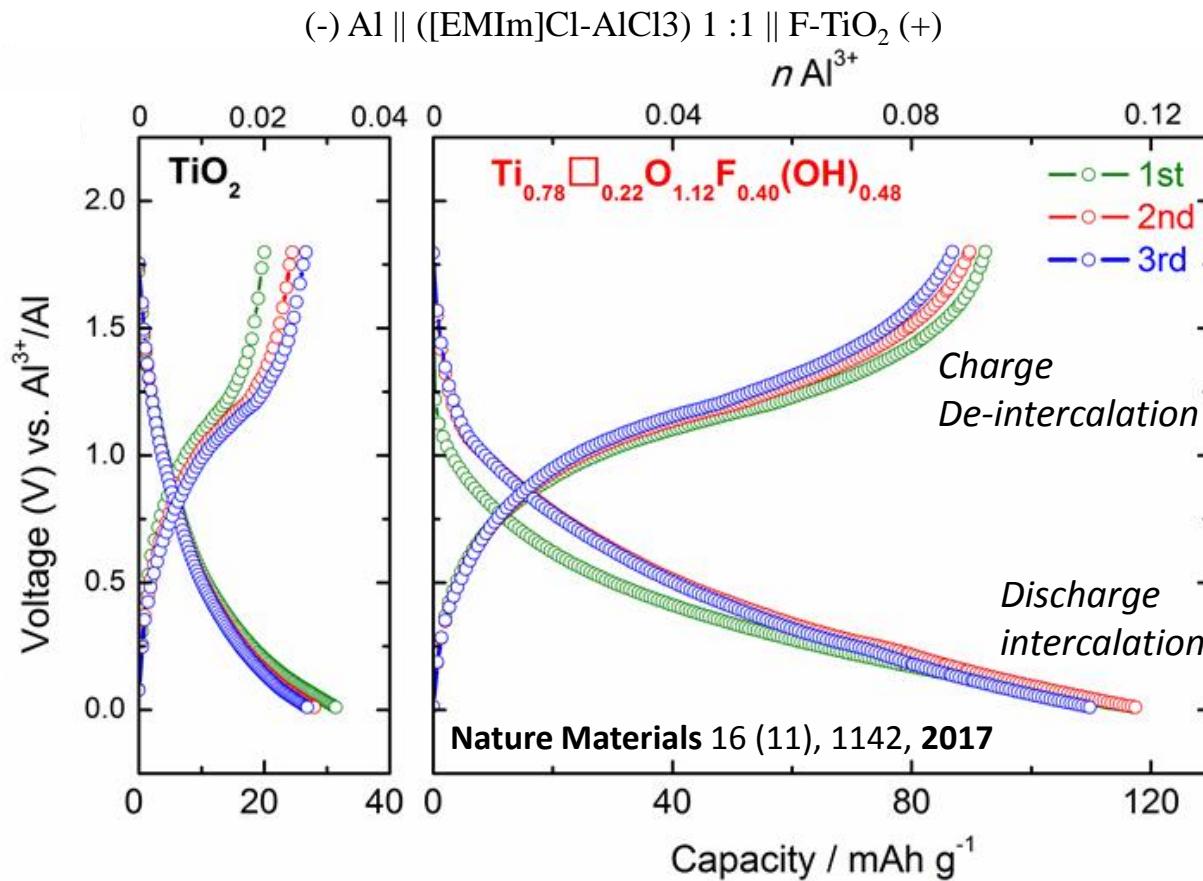
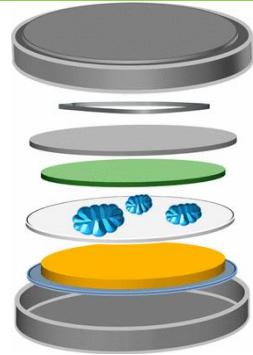
Etude des mécanismes d'intercalation



Nature Materials 16 (11), 1142, 2017

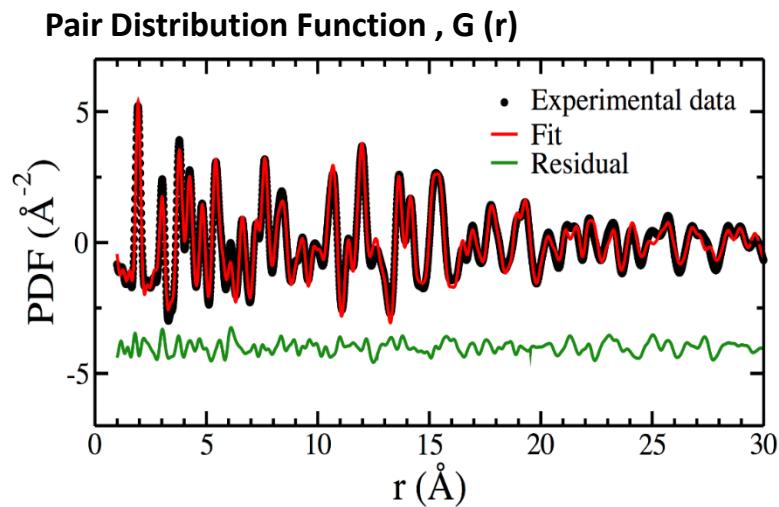
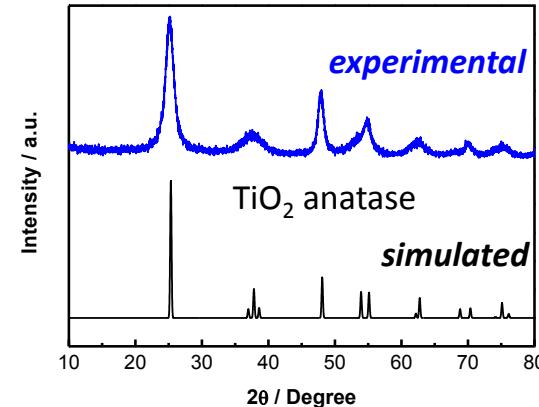
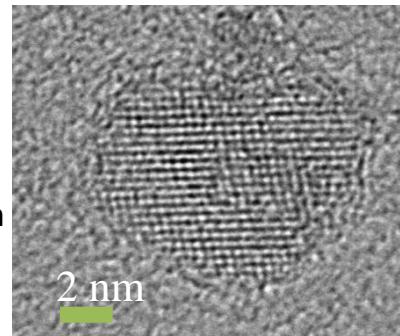
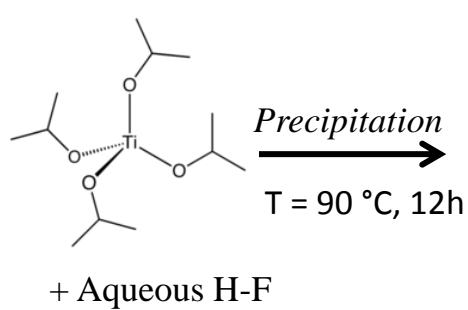


Electrochemical properties vs. Al³⁺/Al



Capacity improvement – Empty vacancies ?

Introduction of titanium vacancies by Hydroxy/fluorination

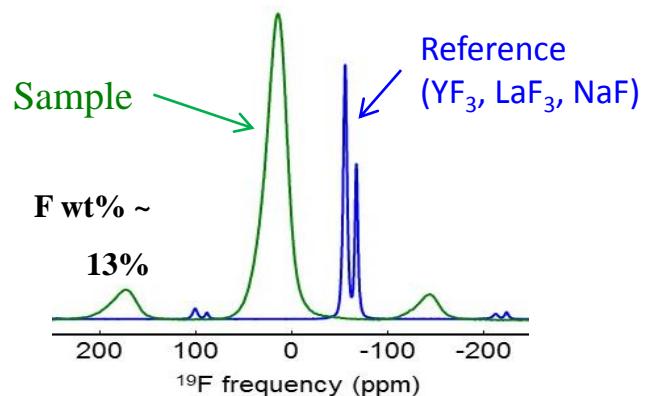


\Rightarrow Structural analysis

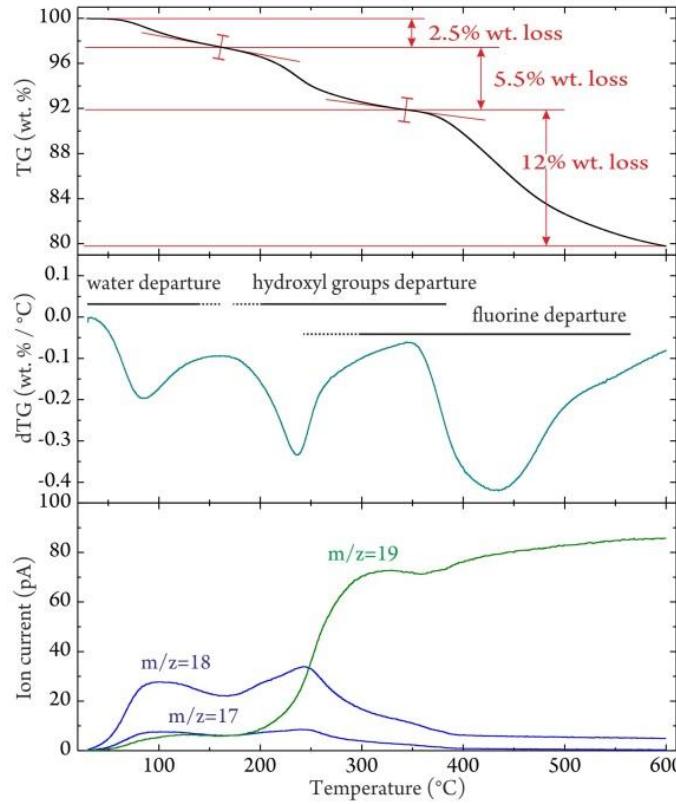
	TiO_2	F-TiO_2
a (\AA)	3.7695(5)	3.784(1)
c (\AA)	9.454(2)	9.448(6)
V (\AA^3)	134.33(4)	135.28(10)
$d_{\text{Ti-O}} (\text{\AA})$	$2*1.972(3)$ $4*1.925(1)$	$2*1.984(6)$ $4*1.929(1)$
Ti (4a) occupancy	1.00(1)	0.74(4)

Determination of the chemical composition

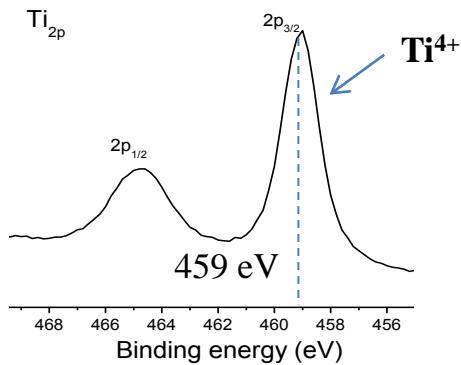
Solid state ^{19}F NMR spectroscopy



ThermoGravimetric Analysis with MS

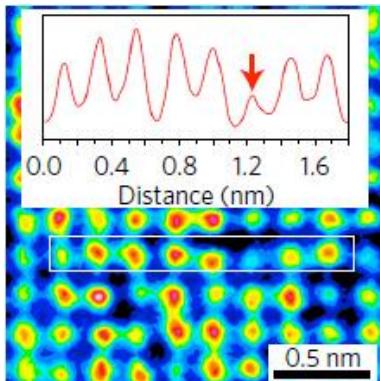
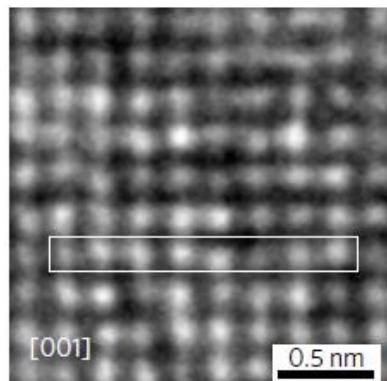


X-ray Photoelectron Spectroscopy

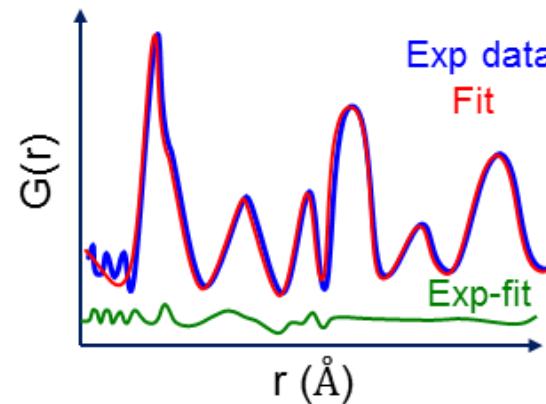


Des matériaux complexes

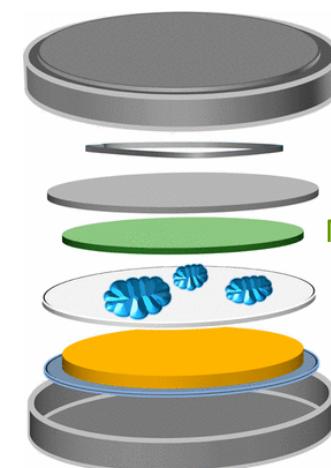
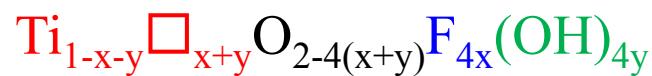
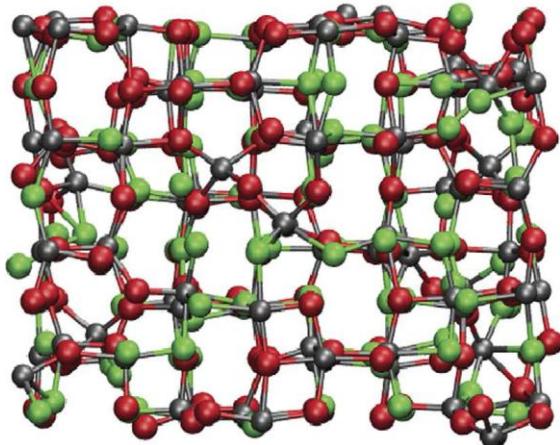
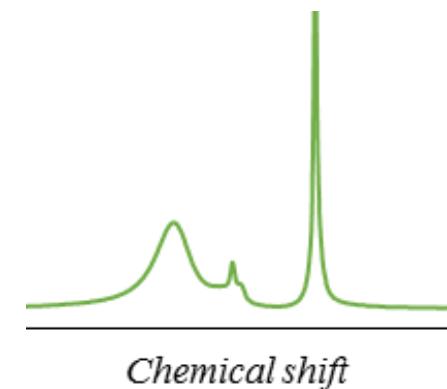
High-resolution TEM



X-ray Pair Distribution Fonction



^{19}F , ^1H Solid-state NMR



Top cap
Spring
Spacer
Anode
Electrolyte
Separator
Cathode
Bottom cap

Nazar et al, Chem mater

